Sulfate attack and the role of internal carbonate on the conditions of thaumasite formation

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Introduction

Concrete deterioration due to the formation of thaumasite caused by sulfate attack has been reported worldwide. In Switzerland the concrete lining of several tunnel structures are affected.

The project aimed to clarify the conditions of thaumasite formation in cementitious systems exposed to various sulfate environments in the laboratory and to determine the role of internal carbonate as part of the chemical reactions. \rightarrow Influence of carbonate (CaCO₃) in the cement

→ Influence of leaching and temperature

→Influence of tricalcium aluminate (C3A) from clinker

Small particles

simulation of leaching (pH)

thermodynamic modeling

separation + analysis of

-> near equilibrium

progressive equilibrium approach PEA



Concept PEA (paste) cement paste experiments mortar experiments Small particles immersed in different Na₂SO₄ solutions formation of thaumasite consequences of thaumasite formation (chemical aspects) variables:



environment cementitious material



Mortar

- Difference in chemistry near surface <-> core [2]
- · Thaumasite formation restricted to surface
- · Initially ettringite and gypsum precipitation;
- thaumasite is last sulfate phase that forms
- Thaumasite not (main) cause for sulfate damage

Test depends on time, Na₂SO₄ concentration, exchange cycles, porosity, temperature, ...





Experimental approach

- Variations in C_3A , $CaCO_3$ content, temperature (8, 20 °C) and comparison leached versus unleached conditions
- Laboratory cements produced with two industrial clinkers
- Comparison of experimental results with thermodynamic modeling
- Cement paste (near equilibrium) and mortar experiments (physical aspects)

- Thaumasite forms only in the presence of high sulfate contents (molar SO₃/Al₂O₃ >3)
- ➤Thaumasite stable at 8 and 20 °C
- Thaumasite formation faster at lower temperatures
- Leaching reduced thaumasite formation slightly [1]

Exposure [days] Fig. 3: Expansion and porosity of mortar samples in ersed in 44 g/l Na₂SO₄ solution

- Limestone influences the hydrate assemblage of hydrated cement before sulfate interaction: -5% -> porosity \downarrow , expansion \downarrow

 - -25% -> porosity \uparrow , expansion \uparrow



Fig. 4: BSE images of different zones in mortar sample immersed in $44g/I Na_2SO_4$ solution after 56 days [2].

[1] T. Schmidt, B. Lothenbach, M. Romer, K. Scrivener, D. Rentsch, R. Figi, A thermodynamic and experimental study of the conditions of thaumasite formation, Cem. Concr. Res. 38 (2008) 337-349.

[2] T. Schmidt, B. Lothenbach, M. Romer, J. Neuenschwander, K. Scrivener, Physical and microstructural aspects of sulfate attack on ordinary and limestone blended Portland cements, Cem. Concr. Res. in press (2009).

Acknowledgements

Literature

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